QP Code: 403008 Reg. No......

Final Year B. Pharm (Ayurveda) Degree Supplementary Examinations September 2022

Pharmaceutical Analysis II

Time: 3 hrs Total Marks: 100

 Answer all questions to the point neatly and legibly
Do not leave any blank pages between answers
Indicate the question number correctly for the answer in the margin space

- Answer all parts of a single question together Leave sufficient space between answers
- Draw diagrams wherever necessary

Essays: (2x10=20)

- Explain in detail about Beers Lamberts Law and discuss about the deviations from Beers Lamberts Law.
- 2. Write the principle and applications of flame emission spectroscopy.

Short notes: (10x5=50)

- 3. Effect of solvent in absorption spectroscopy.
 - 4. Sources of zone broadening in chromatography.
 - 5. Explain the term singlet, triplet and guenching.
 - 6. Explain the applications of NMR spectroscopy.
 - 7. Explain the principle and working of DSC.
 - 8. Explain about hollow cathode lamp in atomic absorption spectroscopy.
 - 9. Electronic transitions in UV spectroscopy.
 - 10. Explain about Gel chromatography.
 - 11. Write about the preparation and development techniques in TLC.
 - 12. Types of ions produced in Mass spectroscopy.

Answer briefly: (10x3=30)

- 13. Polarographic maxima.
- 14. Criteria for IR absorption.
- 15. Filters used in Colorimetry.
- 16. Write any two applications of Amperometric titrations.
- 17. Role of primary and secondary filters in Fluorimeter.
- 18. Chromophore and Auxochrome.
- 19. What are Stokes and Antistoke Fluorescence.
- 20. Define Half wave potential and Diffusion current.
- 21. Define Isobestic point and E1%1CM value.
- 22. Red and Blue shifts.
